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A new class of histamine H₃ receptor antagonists derived from ligand based design

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Abstract—Design and synthesis of highly potent and selective non-imidazole inverse agonists for the histamine H₃ receptor is described. The study validates a new pharmacophore model based on the merging of two previously described models. It also demonstrates that the removal of the basic center potentially interacting with ASP3.32 and common to both models leads to loss of activity, whereas the replacement of the second basic center by an acceptor retains the potency. © 2007 Elsevier Ltd. All rights reserved.

Histamine is a central modulator of physiological processes acting both in the central nervous system (CNS) and the periphery through four distinct receptors (H₁- H_{4}) known to date.^{1,2} Histamine H_{1} and H_{2} receptor antagonists are already used in the clinic to treat allergic asthma and excess gastric acid production, respectively. The newly discovered H₄ receptor seems to have a role in regulating inflammatory responses.³ The H₃ receptor, mainly located in the CNS, is a presynaptic autoreceptor that does not only modulate the production and the release of histamine from histaminergic neurons⁴ but also regulates the release of other neurotransmitters in both the central and peripheral nervous systems. 5-7 Another feature of the H₃ receptor is its high constitutive activity.^{8,9} H₃ antagonists/inverse agonists have been primarily studied in cognitive models for Alzheimer's disease, attention-deficit disorder, and schizophrenia. 10,11 Pharmacological data also suggest a potential role for H₃ antagonists and/or inverse agonists in the control of feeding, appetite, and body weight, and support the role of H₃ receptor in obesity. ^{12,13}

An analysis of known low nanomolar H₃ antagonists published in patents and literature has been made in order to derive a three-dimensional pharmacophore model. Since imidazole-containing compounds could have potential liability toward cytochrome-P450 (CYP)

drug metabolizing enzymes, the imidazole-based H₃ antagonists were discarded. ¹⁴ As it has been shown that basic amine-based molecules have different binding modes than imidazole-based ones, a large part of the modeling information cannot be used. ¹⁵⁻¹⁹ The focus has then been on the basic amine-containing compounds that are postulated to interact with the highly conserved ASP3.32. ¹⁶ During the manual alignment of the molecules, we can observe two distinct pharmacophore distributions (Fig. 1). Each model consists of four pharmacophoric features. Three of the features are shared by both models, namely a distal positive charge, an electron rich position, and a central aromatic ring. The fourth feature can be either a second basic amine exemplified with a potent Johnson & Johnson compound (model 1)²⁰ or another aromatic contained in many Abbott molecules coming from A. Hancock group like A-349821 (model 2), ²¹ but also later ABT-239. ²²

Both types of molecules have been extensively described separately and, at the time of the study, no molecules have been disclosed combining all the features of the combined pharmacophore. Nevertheless, since then, research teams at Johnson & Johnson have published various chemotypes supporting our pharmacophore hypothesis.^{23–25} Our study describes the rational approach followed at Roche to generate new chemotypes validating our assumption. Considering the large overlapping of the pharmacophores, we assume a common binding mode, which will be driven by the interaction between the common basic amine and the conserved aspartate in helix 3 (ASP3.32). Thus, we

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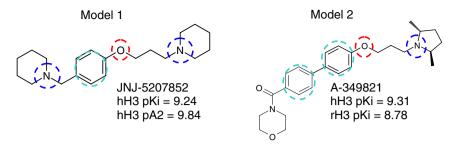


Figure 1. Two pharmacophore models derived from known antagonists of the histamine H₃ receptors. The pharmacophoric features are displayed as dashed circles: blue, positive charge (basic amine); red, electron rich; cyan, aromatic/lipophilic.

designed a new pharmacophore model consisting of five features: two aromatics, two positively charged moieties, and one electron rich feature. We included the electron rich feature because our entire set of reference molecules contains an ether linker at the same position, but the available SAR does not allow us to be sure it is fully mandatory. To provide starting points for the chemistry effort, we use the de novo program Skelgen. ^{26–28} The program automatically builds new molecules from fragment libraries that match the three-dimensional pharmacophoric constraints. It provides hundred of

possible molecules that have been manually clustered. This leads to four templates, which share the same pharmacophores but have different connectivity patterns (Fig. 2). The chemical synthesis has been developed around the connective pattern of template two in order to validate the new pharmacophore.

Synthesis of 4-amino quinazoline of type 1 was better accomplished as indicated in Scheme 1, starting with the O-alkylation of commercially available nitro phenol (3) with benzyl chloride. The reductive amination of the

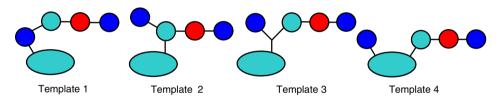


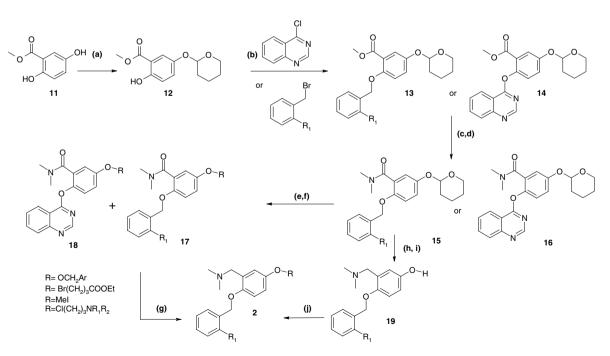
Figure 2. Projection scheme of the various connectivity patterns between the pharmacophore features suggested by Skelgen. Blue, positive charge (basic amine); red, electron rich; cyan, aromatic/lipophilic.

Scheme 1. Reagents and conditions: (a) BnCl (excess), K_2CO_3 , DMF, 110 °C, 4 h, 100%; (b) KMnO₄, acetone/water (1:1), 5 h, 98%; (c) HBTU, dimethylamine hydrochloride, TEA, DMF, 3 h, 89%; (d) BH₃·THF, THF, 70 °C, 8 h, 90% (complex); (e) Pt–C 10%, H₂, EtOAc, 18 h, 80%; (f) POCl₃, 100 °C, microwave, 15 min; (g) **8**, EtOH, DIPEA, reflux, 12 h, 85%; (h) Pd (10% C), EtOH, H₂, rt, overnight, 83%; (i) RX, K_2CO_3 , DMF, 18 h, 60 °C, 40–10% or two-step procedure with (j) Br(CH₂)₃Cl, $C_{52}CO_3$, acetone, rt, 5 h, 93%; (k) Corresponding amine, EtOH, reflux, 5 h, 65–90%.

Table 1. Binding data on human histamine receptors H₁, H₂, H₃ for the compounds of series 1

Compound	X	R	Y	hH ₃ K _i (nM)	hH ₃ % inhibition ^a	hH ₁ % inhibition ^a	hH ₂ % inhibition ^a
ABT-239 ²² JNJ-5207852 ²⁰ 1a	Н	н	N	2.4 1.4	11.6%		
1b	Н	0-	N		31%		
1c	Н		N		28%		
1d	Н	Cl	N		49%		
1e	Н		N	18		-15%	-2%
1f	Н	\sim	N	0.3	_	-2.5%	2.4%
18a	О		О	1021			

 $^{^{\}text{a}}$ Percentage of inhibition measured at 10 $\mu\text{M}.$

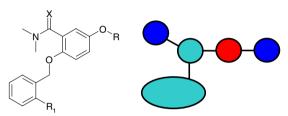


Scheme 2. Reagents and conditions: (a) DHP, PPTS, CH_2Cl_2 , 12 h, 97%; (b) 4-chloro-quinazoline 8, Cs_2CO_3 , DMF, 90 °C, 4 h, 90% or $BrCH_2Ar$, Cs_2CO_3 , 40 °C, acetone, 4 h, 100%; (c) LiOH, THF, H_2O , 40 °C, 18 h, 100%; (d) CDI, $HN(CH_3)_2$ ·HCl, DMF, 60 °C, or HBTU, Et_3N , $HN(CH_3)_2$ ·HCl, DMF, 2 h, 90-70%; (e) PTSOH, MEOH, 60 °C, 8 h, 84%; (f) K_2CO_3 , RX, DMF, 100 °C, 2 h, 90-70%; (g) BH_3 ·THF THF, 90 °C, 4 h, 63%; (h) LiAlH₄, THF, 40 °C, 18 h, 70%; (i) PTSOH, MEOH, 60 °C, 8 h, 69%; (j) RX, K_2CO_3 , DMF, 100 °C, 3 h, 20-70%.

α-nitroaldehyde (4) gave low yields and an alternative route via the reduction of the corresponding amide was followed. The aldehyde (4) was oxidized to the acid to obtain via HBTU coupling the N,N-dimethyl amide (5). Reduction of the amide was performed with borane-THF complex to obtain the amine (6) as a borane complex. After reduction of the nitro group in the presence of the benzyl group with platinum as catalyst, the compound 7 was obtained free of borane. Alkylation of the nitrogen with 4-chloro-quinazoline (8) (obtained from the treatment of commercially available quinazolin-4-ol with POCl₃) was performed in ethanol. The use of DIPEA allows the reaction to proceed rapidly and more cleanly toward 9. Removal of the benzyl group gives the phenol that can react with a range of alkylating agents in the presence of K₂CO₃ to obtain compounds of type 1 described in Table 1. In cases where direct alkylation did not provide high yields, a two-step procedure can be used instead with a first alkylation with 1-bromo-3-chloropropane using Cs₂CO₃ to obtain the *O*-alkylchloride that can be heated with the corresponding amine to prepare 1. Alternative routes with the use of 4-fluorophenol or 2,5-dihydroxybenzal-dehyde gave undesired byproducts and lower yields.

In the case of compounds of type **2** (4-oxo quinazolines or benzyloxy compounds), the most successful synthetic route starts with the regioselective alkylation of the commercially available 2,5-dihydroxy-benzoic acid (**11**) with THP under acidic conditions to obtain **12**.²⁹ Alkylation with 4-chloro-quinazoline (**8**) or with the corresponding alkylbromide with Cs₂CO₃ gives **13** or **14**, respectively. Hydrolysis of the ester using LiOH and amide formation with HNMe₂·HCl using CDI gives

Table 2. Binding data on human histamine receptors H₁, H₂, H₃ for the compounds of series 2



Compound	X	R	R_1	$hH_3 K_i (nM)$	hH ₃ % inhibition ^a	hH ₁ % inhibition ^a	hH ₂ % inhibition ^a
2a	Н	N	Cl	150			
2 b	Н	-	Cl	88			
2c	Н		Cl	135			
2d	Н	\bigvee_{N}	Cl	2.4		_	_
17a	О	F	Cl		33%		
17b	О	\bigvee_{N}	Cl	47		8%	
17c	O	~°	Cl		17%		
17d	O	N O	Cl	1100			
17e	O	N	Cl	1810			
17f	O	N O	OMe	1393	73.4%		
17g	О	N	OMe	12			

 $^{^{\}text{a}}$ Percentage of inhibition measured at 10 $\mu M.$

the amides 15 or 16. In both cases removal of the THP and O-alkylation with an appropriate alkylating agent gave compounds 17 or 18. Reduction of the amide group on 17 or 18 to the corresponding amines using borane–tetrahydrofuran complex gives reasonable yields except on compounds for type 17 when a benzyl or methyl group is present ($R = OCH_2Ar$ and R = Me). An alternative way to obtain the desired amine 2 was through the reduction of compound 15, using LiAlH₄. Removal of the THP group under acidic conditions gives the phenol 19 that can be alkylated with K_2CO_3 and the corresponding halogenated group. The compounds synthesized by this route (Scheme 2) are exemplified in Table 2.

The H₃ activities presented in Tables 1 and 2 are binding data measured by displacement of $[^3H](R)$ - α -methylhistamine (RAMH).³⁰ For selected compounds, human H₁ and H₂ subtype affinity has been measured using [3H]pyrilamine and [3H]tiotidine, respectively (see Tables 1 and 2).³¹ The SAR explored by the molecules shown in these tables validate the new pharmacophore model since compounds 1e, 1f, and 2a-2d carrying 2 basic centers and a lipophilic substituent altogether are highly active at the H₃ receptor. When we remove the distal basic amine in compounds 1a-1d, potentially interacting with ASP3.32, the activity at the H₃ receptor is completely lost. For the molecules 18a, 17b, 17e, and 17g where the proximal basic center has been eliminated by generating the amide, part of activity can be conserved. Especially, compounds 17b and 17g have a K_i of 42 and 12 nM at the H₃ subtype, respectively. From our results, it is likely that both the proximal basic nitrogen and the carbonyl oxygen of the amide bond interact with the same residue. A potential candidate residue for this type of interaction could be Glu5.46, which can be both a hydrogen bond donor (carbonyl, 17g) and acceptor (basic amine, 2d).

From the potent compounds 1e, 1f, 17b tested against human H_1 and H_2 receptors we confirmed that our series are selective for the H_3 subtype. The most promising compound 1f has been further profiled in terms of functionality and species selectivity. It has been tested in a GTP γ S functional assay and was found to be a potent inverse agonist with an EC $_{50}$ of 0.2 nM. Moreover, 1f also displays a high affinity toward the rat H_3 receptor with a K_i of 9.8 nM qualifying the molecule for further in vivo experiments. Preliminary safety data show that 1f does not block the hERG channel.³²

In conclusion, applying a new pharmacophore model, we have discovered highly potent and selective inverse agonists for both human and rat H₃ receptors. In this study, we have exemplified only one of the connection patterns proposed by Skelgen. Based on these promising results, other chemotypes will be explored.

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- 31. Membranes and protocols originated from Euroscreen. For more information, see: characterization of histaminergic receptors *Current Protocols in Pharmacology*, Ed. J. Wiley & Sons, **2006**; Vol. 1, Chapter 19, pp 1–19.
- 32. Compound 1f showed 30% inhibition at 10 μ M concentration at the hERG channel (IC₅₀ > 10 μ M).